

## 20. Photochemical Reactions of Tetrahydroquinoxalin-2(1*H*)-ones and Related Compound

by Takehiko Nishio\*, Masaji Kondo, and Yoshimori Omote

Department of Chemistry, University of Tsukuba, Tsukuba-shi, Ibaraki 305, Japan

(9.X.90)

---

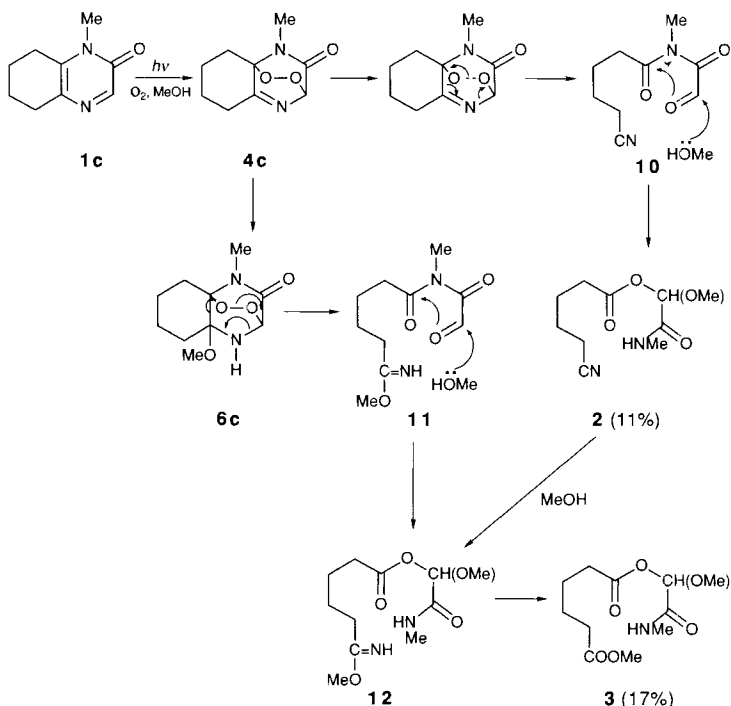
Photochemical behaviors of the pyrazinone derivatives 5,6,7,8-tetrahydroquinoxalin-2(1*H*)-ones **1a–c** and 1,5,6,7,8,9-hexahydro-2*H*-cyclohepta[*b*]pyrazin-2-one **1d** were investigated. Dye-sensitized photo-oxygenation of **1a–c** gave the 1:1 adducts **5a–c** of the corresponding 3,8a-epidioxy-3,5,6,7,8,8a-hexahydroquinoxalin-2(1*H*)-one **4** and H<sub>2</sub>O, whereas **1d** gave 3,9a-epidioxy-1,3,5,6,7,8,9,9a-octahydro-2*H*-cyclohepta[*b*]pyrazin-2-one **4d** (Scheme 2). The different kind of products was interpreted as being the result of the ring strain and steric hindrance of endoperoxides produced from **1a–d** with singlet oxygen. Irradiation of **1a–b** in the presence of alkenes gave tricyclic azetidine derivatives **9** by [2 + 2] cycloaddition of the C=N bond of **1** to the alkene.

---

**1. Introduction.** – The photochemistry of heterocycles possessing an amide functional group such as pyridin-2-ones [1] and uracil derivatives [2] has been extensively studied; however, that of the conjugated cyclohexadienone system containing two N-atoms such as pyrimidinones [3–5], pyrazinones [6] [7], and pyridazinones [8] has received little attention. It is of interest to study the photochemical reactions of cyclic conjugated carbonyl compounds containing N-atoms in view of their relation to nucleoside bases [2]. In earlier work on the photochemical reactivities of cyclic conjugated N–CO systems such as pyrimidinones [3], pyrazinones [6], and quinoxalinones [9], we have demonstrated that the monocyclic pyrazin-2-ones readily reacted with singlet oxygen to form cyclic peroxides [6], while the bicyclic quinoxalin-2-ones underwent [2 + 2] photocycloaddition with alkenes to give tricyclic azetidine derivatives [9]. Therefore, we now examined the photochemical reactions of the bicyclic pyrazinones 5,6,7,8-tetrahydroquinoxalin-2(1*H*)-ones **1a–c** and of the related compound **1d**, in order to understand the generality of these earlier observations.

**2. Results and Discussion.** – *Photochemical Reactions of 1 under Oxygen.* Irradiation of a solution of 1-methyl-5,6,7,8-tetrahydroquinoxalin-2(1*H*)-one (**1c**) in MeOH saturated with O<sub>2</sub> in a Pyrex vessel with a high-pressure mercury lamp for 2 h at room temperature gave the two carbamoylmethyl derivatives **2** and **3** in 11 and 17% yield, respectively (Scheme 1). The reaction appears to proceed through the intermediate transannular peroxides **4c** or **6c** which undergo O–O bond fission, rearrangement, and addition of MeOH to form carbamoylmethyl cyanopentanoate **2** and carbamoylmethyl methyl hexanedioate **3**. The intermediate endoperoxides **4c** or **6c** could not be isolated in pure form; however, the formation of **6c** was confirmed by the <sup>1</sup>H-NMR spectrum (2.98 (*s*, MeN); 3.12 (*s*, MeO); 5.04 ppm (*d*, *J* = 3 Hz, NH)). Ester **3** may be formed by hydrolysis of the methoxyimidoyl group of **12** which derives from **6c** or by methanolysis

Scheme 1



of intermediate **2**. The formation of endoperoxides from **1c** can be interpreted as photo-oxygenation with singlet oxygen in which **1c** acts as self-sensitizer [6c].

Hence dye-sensitized photo-oxygenation of compounds **1a–d** was performed. As expected, irradiation of 5,6,7,8-tetrahydroquinoxalin-2(1*H*)-ones **1a–c** in  $CH_2Cl_2$  in the presence of methylene blue as sensitizer with visible light at room temperature under  $O_2$  gave the 1:1 adducts of the corresponding endoperoxide **4** and  $H_2O$ , *i.e.* 3,8a-epidioxy-octahydro-4a-hydroxyquinoxalin-2(1*H*)-ones **5a–c**, in 41–67% yields after chromatography on silica gel (Scheme 2, Table). Similarly, irradiation of **1a** in MeOH under the same conditions gave 3,8a-epidioxy-octahydro-4a-methoxyquinoxalin-2(1*H*)-one **6a** in 61% yield. The structure of endoperoxides **5a–c** and **6a** was elucidated on the basis of their spectral data and elemental analyses, and their formation can be explained by the instability of the endoperoxides **4a–c**, which is due to the ring strain of fused six-membered rings. The same products **5a–c** were obtained when the dye-sensitized photo-oxygenation of **1a–c** was carried out in the presence of molecular sieves to remove traces of  $H_2O$  in the solvent: the formation of the endoperoxides **4a–c** was supported by the fact that the reaction mixtures showed a new IR absorption at *ca.*  $1630\text{ cm}^{-1}$  ( $C=N$ ), and thus  $H_2O$  addition probably occurred during chromatography on silica gel. On the other hand, dye-sensitized photo-oxygenation of 1,5,6,7,8,9-hexahydro-2*H*-cyclohepta[*b*]pyrazin-2-one **1d** in both  $CH_2Cl_2$  and MeOH under the same condition yielded 3,9a-epidioxy-1,3,5,6,7,8,9,9a-octahydro-2*H*-cyclohepta[*b*]pyrazin-2-one **4d** in 75% yield and no

Scheme 2

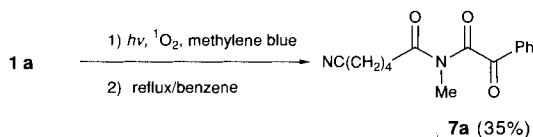
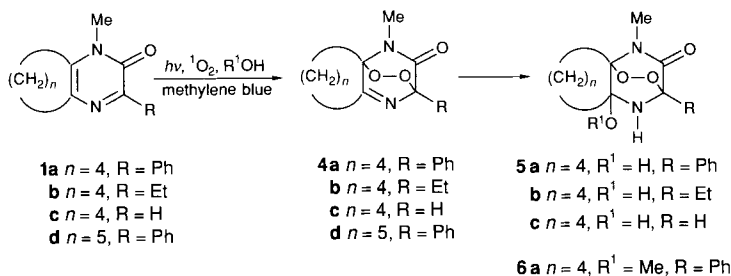


Table. Yield of Endoperoxides 4-6

Starting material	Solvent	Products (yield [%] <sup>a</sup> )
<b>1a</b> $n = 4$ $R = \text{Ph}$	$\text{CH}_2\text{Cl}_2$	<b>5a</b> (67)
<b>1a</b> $n = 4$ $R = \text{Ph}$	$\text{MeOH}$	<b>6a</b> (61)
<b>1b</b> $n = 4$ $R = \text{Et}$	$\text{CH}_2\text{Cl}_2$	<b>5b</b> (46)
<b>1c</b> $n = 4$ $R = \text{H}$	$\text{CH}_2\text{Cl}_2$	<b>5c</b> (41)
<b>1d</b> $n = 5$ $R = \text{Ph}$	$\text{CH}_2\text{Cl}_2$	<b>4d</b> (75)
<b>1d</b> $n = 5$ $R = \text{Ph}$	$\text{MeOH}$	<b>4d</b> (71)

<sup>a</sup>) Isolated yield.

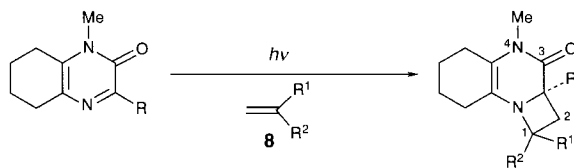
$\text{MeOH}$  adduct. This is compatible with the increased stability of **4d** due to less ring strain as a result of the flexibility of the seven-membered ring.

Unsymmetrical cyano-imide **7** was obtained in a one-pot reaction of **1a** [6c]. Dye-sensitized photo-oxygenation of **1a** in benzene, followed by thermal decomposition gave **7** in 35% yield, which corresponds to the intermediate **10** shown in Scheme 1.

[2 + 2] Photocycloaddition of the Tetrahydroquinoxalin-2(1H)-ones **1a, b** to Alkenes **8**. Recently, we reported that upon irradiation, quinoxalin-2-ones and benzoxazin-2-ones undergo [2 + 2] cycloaddition with electron-poor alkenes [9a,b,d] or arylalkenes [9c] to yield tricyclic azetidines. However, monocyclic pyrazin-2-ones do not react under these conditions with alkenes. Therefore, it is of interest to investigate the photocycloaddition of **1** to alkenes **8** which should reveal the effect of the ring condensed to pyrazin-2-one.

Irradiation of a solution of 5,6,7,8-tetrahydro-3-phenylquinoxalin-2(1H)-one **1a** and excess methacrylonitrile in benzene in a Pyrex vessel with a high-pressure mercury lamp under Ar at room temperature gave the 1:1 cycloadduct **9a** in 40% yield (Scheme 3). The formation of azetidine **9a** was quenched by *trans*-stilbene as a triplet quencher. The structure of **9a** was established by spectral data and elemental analysis. Azetidine **9a** converted back to the starting quinoxalin-2-one **1a** when **9a** was heated at  $80^\circ/10^{-3}$  Torr.

Scheme 3



**1a** R = Ph  
**b** R = Et

**9a** R = Ph, R<sup>1</sup> = Me, R<sup>2</sup> = CN; yield 40%<sup>a)</sup>  
**b** R = Ph, R<sup>1</sup> = H, R<sup>2</sup> = CN; yield 8%<sup>a)</sup>  
**c** R = Ph, R<sup>1</sup> = Me, R<sup>2</sup> = COOMe; yield 16%<sup>a) b)</sup>  
**d** R = R<sup>2</sup> = Ph, R<sup>1</sup> = H; yield 36%<sup>a)</sup>  
**e** R = R<sup>2</sup> = Ph, R<sup>1</sup> = Me; yield 38%<sup>a)</sup>  
**f** R = R<sup>1</sup> = R<sup>2</sup> = Ph; yield 49%<sup>a)</sup>  
**g** R = Et, R<sup>1</sup> = Me, R<sup>2</sup> = CN; yield 25%<sup>a)</sup>  
**h** R = Et, R<sup>1</sup> = Me, R<sup>2</sup> = Ph; yield 10%<sup>a)</sup>

a) Isolated yield. b) Another stereoisomer was present to 18%.

The <sup>1</sup>H-NMR spectrum of **9a** showed newly formed *AB* signals at  $\delta(\text{H})$  2.49 ( $J = 11.7$  Hz, 1H) and 3.57 ( $J = 11.7$  Hz, 1H) assignable to a CH<sub>2</sub> group of the azetidine ring. Furthermore, in the <sup>13</sup>C-NMR spectrum, signals at  $\delta(\text{C})$  45.3 (*t*), 57.1 (*s*), and 64.9 (*s*) due to azetidine-ring C-atoms newly appeared, and the C=N signal of **1a** at  $\delta(\text{C})$  148.2 (*s*, C(3)) had disappeared. The regiochemistry of the 1:1 cycloaddition to **9a** was suggested by the chemical shift of the CH<sub>2</sub> group of the azetidine ring [9].

Similarly, upon irradiation, the 5,6,7,8-tetrahydroquinoxalin-2(1*H*)-ones **1a–b** added regioselectively to electron-poor alkenes such as acrylonitrile, methacrylonitrile, and methyl methacrylate or to aryl alkenes such as styrene,  $\alpha$ -methylstyrene, and 1,1-diphenylethylene to give azetidine derivatives **9b–h**. In the reaction of **1a** with methyl methacrylate, two stereoisomers were produced. Based on comparison with NMR data of known azetidine derivatives [9], we tentatively assigned structure **9c** to the less abundant one.

Quinoxalin-2(1*H*)-one **1a** did not undergo photocycloaddition to 1,2-disubstituted alkenes such as crotononitrile and methyl crotonate. Irradiation of **1c** in the presence of alkenes gave several unseparable mixtures. Upon irradiation, the 2*H*-cyclohepta[*b*]-pyrazin-2-one **1d** did not undergo cycloaddition to alkenes, and **1d** was recovered unchanged.

The regiospecificity and the lack of stereospecificity in the [2 + 2] photocycloadditions described above suggest that the formation of azetidine derivatives may arise from initial interaction between the quinoxalin-2(1*H*)-one **1** in the triplet state and alkene to give an excited complex as exciplex [10] which subsequently gives a 1,4-biradical intermediate which cyclizes to the azetidine. Based upon the above results and previous observations [9], it appears that ring constraint<sup>1)</sup> and additional conjugation with a electron-withdrawing carbonyl or aryl function at the N- or C-atom [11] are necessary to achieve [2 + 2] photocycloadditions of C=N and C=C bonds.

<sup>1)</sup> Photocycloadditions of simple imines are very rare. Ring constraint due to the incorporation of the C=N function in a ring system, such as six- or five-membered ring, blocks excited-state deactivation by  $\pi$ -bond rotation or N-inversion [11].

## Experimental Part

1. *General.* Silica gel (*Merck 60* and *Wakogel C-300* for flash chromatography) was used for column chromatography. M.p.: uncorrected. UV spectra: *Shimadzu-UV-365 spectrophotometer*. IR spectra: *Jasco-IRA-1 spectrophotometer*.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra: *Jeol-FX-100* (100 MHz) spectrometer, in  $\text{CDCl}_3$  as solvent, using TMS as an internal standard (exceptions noted in parenthesis). Mass spectra: *Hitachi-M-80 spectrometer*.

2. *Starting Materials.* The 5,6,7,8-tetrahydroquinoxalin-2(1H)-ones **1a–c** and 1,5,6,7,8,9-hexahydro-2H-cyclohepta[b]pyrazin-2-one (**1d**) were prepared by methylation of the corresponding hydroquinoxalin-2-ol and hydrocyclohepta[b]pyrazin-2-ol which were synthesized according to the method previously described [12]. To a stirred soln. of 5,6,7,8-tetrahydroquinoxalin-2-ol or 6,7,8,9-tetrahydro-5H-cyclohepta[b]pyrazin-2-ol (1 mmol) and NaOMe (from Na (1.1 mmol) and MeOH (5 ml)) in MeOH (15 ml) was added dropwise  $\text{Me}_2\text{SO}_4$  (1.1 mmol) at r.t. The mixture was refluxed for 1 h, then poured into 10% HCl soln. and extracted with  $\text{CH}_2\text{Cl}_2$ . The extract was washed with 10%  $\text{NaHCO}_3$  soln. and  $\text{H}_2\text{O}$ , dried ( $\text{MgSO}_4$ ), and evaporated and the residue chromatographed (silica gel, benzene/AcOEt 10:1 to 2:1 or  $\text{CHCl}_3$ /acetone/EtOH 100:5:1).

5,6,7,8-Tetrahydro-1-methyl-3-phenylquinoxalin-2(1H)-one (**1a**). Yield: 72%. M.p. 136.5–138°. UV (EtOH): 230 (6900), 255 (8300), 364 (14700). IR (KBr): 1640 (C=O).  $^1\text{H}$ -NMR: 1.70–1.90 (m, 4H); 2.50–2.85 (m, 4H); 3.48 (s, 3H); 7.26–7.46 (m, 3H); 8.19–8.32 (m, 2H).  $^{13}\text{C}$ -NMR: 22.0 (t); 26.5 (t); 30.1 (t); 30.1 (q); 127.7 (d); 128.6 (d); 128.9 (d); 130.9 (s); 134.4 (s); 148.2 (s); 155.5 (s). Anal. calc. for  $\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}$ : C 74.97, H 6.71, N 11.65; found: C 74.81, H 6.72, N 11.62.

3-Ethyl-5,6,7,8-tetrahydro-1-methylquinoxalin-2(1H)-one (**1b**). Yield: 18%. M.p. 59–60°. UV (EtOH): 231 (4500), 332 (4150). IR (KBr): 1640 (C=O).  $^1\text{H}$ -NMR: 1.23 (t, 3H); 1.65–1.95 (m, 4H); 2.50–2.77 (m, 4H); 2.80 (q, 2H); 3.47 (s, 3H).  $^{13}\text{C}$ -NMR: 11.2 (q); 22.0 (t); 26.1 (t); 26.9 (t); 29.7 (q); 29.9 (t); 129.8 (s); 132.4 (s); 155.8 (s); 156.8 (s). Anal. calc. for  $\text{C}_{11}\text{H}_{16}\text{N}_2\text{O}$ : C 68.71, H 8.38, N 14.57; found: C 68.19, H 8.29, N 14.28.

5,6,7,8-Tetrahydro-1-methylquinoxalin-2(1H)-one (**1c**). Yield: 40%. M.p. 126.5–128°. UV (EtOH): 231 (8300), 340 (6900). IR (KBr): 1650 (C=O).  $^1\text{H}$ -NMR: 1.66–1.95 (m, 4H); 2.50–2.80 (m, 4H); 3.48 (s, 3H); 8.01 (s, 1H).  $^{13}\text{C}$ -NMR: 21.7 (t); 21.8 (t); 26.1 (t); 29.5 (q); 29.6 (t); 131.4 (s); 135.0 (s); 144.2 (s); 156.3 (s). Anal. calc. for  $\text{C}_9\text{H}_{12}\text{N}_2\text{O}$ : C 65.83, H 7.36, N 17.06; found: C 65.58, H 7.34, N 16.96.

1,5,6,7,8,9-Hexahydro-1-methyl-3-phenyl-2H-cyclohepta[b]pyrazin-2-one (**1d**). Yield: 73%. M.p. 104–105°. UV (EtOH): 230 (8200), 257 (7900), 369 (14400). IR (KBr): 1630 (C=O).  $^1\text{H}$ -NMR: 1.56–2.00 (m, 6H); 2.75–3.05 (m, 4H); 3.62 (s, 3H); 7.26–7.55 (m, 3H); 8.20–8.37 (m, 2H).  $^{13}\text{C}$ -NMR: 24.8 (t); 26.1 (t); 29.6 (t); 31.3 (q); 31.6 (t); 36.0 (t); 127.8 (d); 128.6 (d); 128.9 (d); 136.4 (s); 136.7 (s); 140.3 (s); 146.8 (s); 155.3 (s). Anal. calc. for  $\text{C}_{16}\text{H}_{18}\text{N}_2\text{O}$ : C 75.56, H 7.13, N 11.01; found: C 75.40, H 7.13, N 10.95.

3. *Photoreaction of 1c.* A soln. of **1c** (200 mg) in MeOH (50 ml) was irradiated in a *Pyrex* vessel with a Hg high-pressure lamp (450 W) under  $\text{O}_2$  for 5 h at r.t. After evaporation, the residual oil was chromatographed (silica-gel column,  $\text{CHCl}_3$ /acetone/EtOH 100:10:2): **2** and **3**. These products could not be completely separated in anal. pure form by column chromatography on silica gel.

*Methoxy(methylcarbamoyl)methyl 5-Cyanopentanoate (2).* Oil. IR ( $\text{CHCl}_3$ ): 3430, 2225, 1740, 1685.  $^1\text{H}$ -NMR: 1.65–1.94 (m, 4H); 2.30–2.57 (m, 4H); 2.85 (d,  $J = 4.9$ , 3H); 3.53 (s, 3H); 5.91 (s, 1H); 6.69 (br. s, 1H, exchangeable with  $\text{D}_2\text{O}$ ).  $^{13}\text{C}$ -NMR: 17.0 (t); 23.7 (t); 24.7 (t); 26.0 (q); 33.1 (t); 57.9 (q); 93.6 (d); 119.4 (s); 166.2 (s); 172.1 (s). CI-MS: 229 ( $[M + 1]^+$ ).

*Methoxy(methylcarbamoyl)methyl Methyl Hexanedioate (3).* Oil. IR ( $\text{CHCl}_3$ ): 3430, 1730, 1685.  $^1\text{H}$ -NMR: 1.63–1.77 (m, 4H); 2.24–2.53 (m, 4H); 2.86 (d,  $J = 5.4$ , 3H); 3.53 (s, 3H); 3.67 (s, 3H); 5.92 (s, 1H); 6.67 (br. s, 1H, exchangeable with  $\text{D}_2\text{O}$ ).  $^{13}\text{C}$ -NMR: 24.0 (t); 24.1 (t); 25.8 (q); 33.5 (t); 33.6 (t); 51.4 (q); 57.4 (q); 93.3 (d); 166.1 (s); 172.4 (s); 173.5 (s). CI-MS: 262 ( $[M + 1]^+$ ).

4. *Dye-Sensitized Photo-oxygenation of 1.* An oxygenated soln. of **1** (200 mg) in  $\text{CH}_2\text{Cl}_2$  or MeOH (70 ml) in the presence of methylene blue as a sensitizer was irradiated in a *Pyrex* tube with a halogen lamp for 1 h at r.t. The sensitizer was filtered off through silica gel, the solvent evaporated, and the residue chromatographed (silica-gel column, benzene/AcOEt 4:1 to 1:9).

3,9a-Epidioxy-1,3,5,6,7,8,9,9a-octahydro-1-methyl-3-phenyl-2H-cyclohepta[b]pyrazin-2-one (**4d**). M.p. 131.4–133°. IR (KBr): 1700, 1625.  $^1\text{H}$ -NMR: 1.59–2.00 (m, 6H); 2.05–2.32 (m, 2H); 2.58–2.90 (m, 2H); 2.99 (s, 3H); 7.37–7.49 (m, 3H); 7.77–7.91 (m, 2H).  $^{13}\text{C}$ -NMR: 23.4 (t); 26.4 (q); 26.7 (t); 29.2 (t); 30.9 (t); 37.6 (t); 89.1 (s); 91.2 (s); 128.0 (d); 128.8 (d); 129.9 (d); 130.9 (d); 169.9 (s); 182.6 (s). Anal. calc. for  $\text{C}_{16}\text{H}_{18}\text{N}_2\text{O}_3$ : C 67.11, H 6.33, N 9.78; found: C 66.94, H 6.30, N 9.79.

3,8a-Epidioxy-3,4,4a,5,6,7,8,8a-octahydro-4a-hydroxy-1-methyl-3-phenylquinoxalin-2(1H)-one (**5a**). M.p. 130.5–131.5° (dec.). IR (KBr): 3440, 3320.  $^1\text{H}$ -NMR: 1.55–1.80 (m, 4H); 1.80–2.12 (m, 4H); 2.69 (s, 1H, exchangeable with  $\text{D}_2\text{O}$ ); 3.01 (s, 3H); 3.35 (br. s, 1H, exchangeable with  $\text{D}_2\text{O}$ ); 7.34–7.60 (m, 5H).  $^{13}\text{C}$ -NMR:

20.3 (t); 20.5 (t); 26.5 (t); 26.8 (q); 35.6 (t); 79.7 (s); 90.2 (s); 90.9 (s); 126.6 (d); 128.2 (d); 129.2 (d); 132.0 (s); 166.3 (s). Anal. calc. for  $C_{15}H_{18}N_2O_4$ : C 60.06, H 6.25, N 9.65; found: C 61.86, H 6.27, N 9.55. CI-MS: 291 ( $[M + 1]^+$ ).

**3,8a-Epidioxy-3-ethyl-3,4,4a,5,6,7,8,8a-octahydro-4a-hydroxy-1-methylquinoxalin-2(1H)-one (5b)**. M.p. 144–145° (dec.). IR (KBr): 3420, 3340, 1710, 1690.  $^1H$ -NMR ( $(D_6)$ DMSO): 0.94 (t, 3H); 1.30–1.97 (m, 10H); 2.85 (s, 3H); 4.43 (s, 1H); 5.24 (s, 1H).  $^{13}C$ -NMR ( $(D_6)$ DMSO): 7.87 (q); 20.4 (t); 23.5 (t); 26.0 (t); 26.0 (q); 35.5 (t); 78.8 (s); 89.5 (s); 90.0 (s); 166.1 (s). Anal. calc. for  $C_{11}H_{18}N_2O_4$ : C 54.53, H 7.48, N 11.56; found: C 54.24, H 7.44, N 11.44.

**3,8a-Epidioxy-3,4,4a,5,6,7,8,8a-octahydro-4a-hydroxy-1-methylquinoxalin-2(1H)-one (5c)**. IR (KBr): 3370, 3300, 1710.  $^1H$ -NMR ( $(D_6)$ DMSO): 1.14–2.15 (m, 8H); 2.85 (s, 3H); 4.86 (d,  $J = 4.4$ , 1H, exchangeable with  $D_2O$ ); 5.00 (d,  $J = 4.4$ , 1H); 5.37 (s, 1H, exchangeable with  $D_2O$ ).  $^{13}C$ -NMR ( $(D_6)$ DMSO): 20.1 (t); 20.4 (t); 25.8 (t); 25.8 (q); 35.4 (t); 78.6 (s); 84.9 (d); 90.6 (s); 165.2 (s). CI-MS: 215 ( $[M + 1]^+$ ).

**3,8a-Epidioxy-3,4,4a,5,6,7,8,8a-octahydro-4a-methoxy-1-methyl-3-phenylquinoxalin-2(1H)-one (6a)**. M.p. 147–148° (dec.). IR (KBr): 3280, 1695.  $^1H$ -NMR: 1.38–1.98 (m, 6H); 1.98–2.50 (m, 2H); 2.80 (br. s, 1H, exchangeable with  $D_2O$ ); 3.02 (s, 3H); 3.22 (s, 3H); 7.28–7.60 (m, 5H).  $^{13}C$ -NMR: 20.2 (t); 20.4 (t); 26.5 (t); 26.9 (q); 30.5 (t); 46.9 (q); 83.6 (s); 90.1 (s); 90.3 (s); 126.7 (d); 128.2 (d); 129.5 (d); 132.8 (s); 166.0 (s). Anal. calc. for  $C_{16}H_{20}N_2O_4$ : C 63.14, H 6.62, N 9.20; found: C 63.01, H 6.63, N 9.10. CI-MS: 305 ( $[M + 1]^+$ ).

**5. Imide 7 by Dye-Sensitized Photo-oxygenation of 1a Followed by Thermolysis**. An oxygenated soln. of **1a** (100 mg) in  $CH_2Cl_2$  (70 ml) in the presence of methylene blue (ca. 2 mg) as sensitizer was irradiated for 1 h under the same conditions as described above. The sensitizer was filtered off, the solvent evaporated, and the mixture dissolved in benzene (30 ml) and refluxed for 2 h. Usual workup gave 5-cyano-N-methyl-N-(2-oxo-2-phenylethanoyl)pentanamide (**7**; 40 mg, 35%). IR ( $CHCl_3$ ): 2225, 1710, 1685.  $^1H$ -NMR: 1.56–1.88 (m, 4H); 2.19–2.38 (m, 2H); 2.52–2.72 (m, 2H); 3.33 (s, 3H); 7.32–7.62 (m, 3H); 7.74–7.87 (m, 2H).  $^{13}C$ -NMR: 16.8 (t); 22.5 (t); 24.4 (t); 29.7 (q); 34.3 (t); 119.2 (s); 128.7 (d); 128.9 (d); 132.5 (s); 133.9 (d); 169.9 (s); 174.7 (s); 187.0 (s). CI-MS: 273 ( $[M + 1]^+$ ).

**6. General Procedure for the Photocycloaddition of 1a,b to Alkenes**. A soln. of **1a,b** (200 mg) and alkene (ca. 2 ml) in benzene (70 ml) was irradiated in a Pyrex vessel with a Hg high-pressure lamp under Ar for 2–12 h at r.t. After removal of the solvent, the residual oil was chromatographed (silica-gel column, benzene/AcOEt 10:1 or benzene only) to give the azetidines derivatives **9**.

**2,2a,3,4,5,6,7,8-Octahydro-1,4-dimethyl-3-oxo-2a-phenyl-1H-azeto[1,2-a]quinoxaline-1-carbonitrile (9a)**. M.p. 141–142°. IR (KBr): 2220, 1670, 1650.  $^1H$ -NMR: 1.23–2.36 (m, 8H); 1.69 (s, 3H); 2.49, 3.57 (AB,  $J = 11.7$ , 2H); 3.09 (s, 3H); 7.22–7.60 (m, 5H).  $^{13}C$ -NMR: 21.6 (t); 22.4 (t); 25.6 (t); 27.7 (t); 28.3 (q); 28.7 (q); 45.3 (t); 57.1 (s); 64.9 (s); 119.4 (s); 119.7 (s); 124.9 (d); 125.6 (s); 127.7 (d); 128.4 (d); 141.4 (s); 166.6 (s). CI-MS: 308 ( $[M + 1]^+$ ). Anal. calc. for  $C_{19}H_{21}N_3O$ : C 74.23, H 6.88, N 13.66; found: C 73.89, H 6.83, N 13.64.

**2,2a,3,4,5,6,7,8-Octahydro-4-methyl-3-oxo-2a-phenyl-1H-azeto[1,2-a]quinoxaline-1-carbonitrile (9b)**. IR (KBr): 2240, 1670, 1645.  $^1H$ -NMR: 1.20–2.45 (m, 8H); 3.10 (s, 3H); 2.90 (A of ABX,  $J = 8.3$ , 11.7, 1H); 3.37 (B of ABX,  $J = 2.9$ , 11.7, 1H); 4.49 (X of ABX,  $J = 2.9$ , 8.3, 1H); 7.22–7.59 (m, 5H).  $^{13}C$ -NMR: 21.4 (t); 22.5 (t); 25.5 (t); 26.1 (t); 28.7 (q); 37.3 (t); 48.1 (d); 68.4 (s); 117.1 (s); 119.4 (s); 124.9 (d); 126.0 (s); 127.9 (d); 128.5 (d); 141.0 (s); 166.4 (s). CI-MS: 294 ( $[M + 1]^+$ ).

**Methyl 2,2a,3,4,5,6,7,8-Octahydro-1,4-dimethyl-3-oxo-2a-phenyl-1H-azeto[1,2-a]quinoxaline-1-carboxylate (9c)**. The 2 stereoisomers present could not be completely separated. Isomer **9c** (16%). IR ( $CHCl_3$ ): 1730, 1630.  $^1H$ -NMR: 1.08–2.40 (m, 8H); 1.51 (s, 3H); 2.87, 3.08 (AB,  $J = 12.0$ , 2H); 3.07 (s, 3H); 3.75 (s, 3H); 7.21–7.60 (m, 5H).  $^{13}C$ -NMR: 20.8 (q); 21.8 (t); 22.7 (t); 25.6 (t); 27.6 (t); 28.5 (q); 43.7 (t); 52.1 (q); 64.1 (s); 64.8 (s); 120.3 (s); 122.9 (s); 125.2 (d); 127.2 (d); 128.3 (d); 142.1 (s); 167.6 (s); 174.7 (s). CI-MS: 341 ( $[M + 1]^+$ ). Stereoisomer (18%). IR ( $CDCl_3$ ): 1720, 1645.  $^1H$ -NMR: 1.20–2.23 (m, 8H); 1.57 (s, 3H); 2.30, 3.61 (AB,  $J = 12.0$ , 2H); 3.02 (s, 3H); 3.72 (s, 3H); 7.19–7.40 (m, 5H). CI-MS: 341 ( $[M + 1]^+$ ).

**2a,4,5,6,7,8-Hexahydro-4-methyl-1,2a-diphenyl-1H-azeto[1,2-a]quinoxalin-3(2H)-one (9d)**. IR ( $CHCl_3$ ): 1635.  $^1H$ -NMR: 1.17–2.35 (m, 8H); 2.36 (A of ABX,  $J = 7.8$ , 10.8, 1H); 3.08 (s, 3H); 4.76 (B of ABX,  $J = 7.8$ , 11.2, 1H); 4.76 (X of ABX,  $J = 7.8$ , 1H); 7.18–7.70 (m, 10H).  $^{13}C$ -NMR: 21.8 (t); 22.7 (t); 25.5 (t); 27.0 (t); 28.5 (q); 43.7 (t); 64.5 (d); 65.1 (s); 120.9 (d); 124.2 (s); 125.3 (d); 126.6 (d); 127.1 (d); 127.5 (d); 127.7 (d); 128.0 (d); 128.3 (d); 142.9 (s); 143.5 (s); 167.8 (s). CI-MS: 345 ( $[M + 1]^+$ ).

**2a,4,5,6,7,8-Hexahydro-1,4-dimethyl-1,2a-diphenyl-1H-azeto[1,2-a]quinoxalin-3(2H)-one (9e)**. IR (KBr): 1670, 1645.  $^1H$ -NMR: 1.18–2.37 (m, 8H); 1.70 (s, 3H); 2.90, 3.23 (AB,  $J = 11.2$ , 2H); 3.09 (s, 3H); 7.05–7.68 (m, 10H).  $^{13}C$ -NMR: 21.7 (t); 22.7 (t); 23.4 (t); 25.6 (t); 27.9 (q); 28.5 (q); 49.7 (t); 63.8 (s); 65.2 (s); 121.1 (s); 121.5 (s); 125.1 (d); 125.3 (d); 126.9 (d); 128.2 (d); 143.1 (s); 148.3 (s); 168.0 (s). CI-MS: 359 ( $[M + 1]^+$ ).

*2a,4,5,6,7,8-Hexahydro-4-methyl-1,1,2a-triphenyl-1H-azeto[1,2-a]quinoxalin-3(2H)-one (9f)*. M.p. 143–144°. IR (CHCl<sub>3</sub>): 1640. <sup>1</sup>H-NMR: 1.05–2.30 (*m*, 8H); 2.68 (*s*, 3H); 2.70, 4.20 (*AB*, *J* = 12.2, 2H); 7.05–7.78 (*m*, 15H). <sup>13</sup>C-NMR: 21.5 (*t*); 22.4 (*t*); 25.2 (*t*); 28.0 (*q*); 28.9 (*t*); 47.4 (*t*); 64.5 (*s*); 73.7 (*s*); 121.2 (*s*); 124.0 (*s*); 125.3 (*d*); 126.4 (*d*); 127.0 (*d*); 127.4 (*d*); 127.5 (*d*); 127.8 (*d*); 128.2 (*d*); 140.2 (*s*); 142.7 (*s*); 148.2 (*s*); 168.0 (*s*). CI-MS: 421 ([*M* + 1]<sup>+</sup>).

*2a-Ethyl-2,2a,3,4,5,6,7,8-octahydro-1,4-dimethyl-3-oxo-1H-azeto[1,2-a]quinoxaline-1-carbonitrile (9g)*. IR (CHCl<sub>3</sub>): 2240, 1640. <sup>1</sup>H-NMR: 0.91 (*t*, 3H); 1.26–2.10 (*m*, 8H); 1.63 (*s*, 3H); 2.16–2.38 (*s*, 2H); 2.33, 3.07 (*AB*, *J* = 11.7, 2H); 3.12 (*s*, 3H). <sup>13</sup>C-NMR: 7.2 (*q*); 21.3 (*t*); 22.3 (*t*); 25.4 (*t*); 27.4 (*t*); 28.1 (*t*); 28.1 (*q*); 32.1 (*q*); 41.2 (*t*); 57.0 (*s*); 63.9 (*s*); 119.4 (*s*); 120.0 (*s*); 124.0 (*s*); 168.2 (*s*). CI-MS: 260 ([*M* + 1]<sup>+</sup>).

*2a-Ethyl-2a,4,5,6,7,8-hexahydro-1,4-dimethyl-1-phenyl-1H-azeto[1,2-a]quinoxalin-3(2H)-one (9h)*. IR (CHCl<sub>3</sub>): 1630. <sup>1</sup>H-NMR: 0.93 (*t*, 3H); 1.08–2.10 (*m*, 8H); 2.10–2.39 (*m*, 3H); 2.81 (*d*, *J* = 11.2, 1H); 3.15 (*s*, 3H); 7.05–7.45 (*m*, 3H); 7.50–7.70 (*m*, 2H). <sup>13</sup>C-NMR: 8.1 (*q*); 21.7 (*t*); 22.8 (*t*); 24.0 (*q*); 25.7 (*t*); 28.2 (*t*); 28.2 (*q*); 33.2 (*t*); 46.6 (*t*); 62.8 (*s*); 64.8 (*s*); 120.5 (*s*); 121.8 (*s*); 124.8 (*d*); 126.3 (*d*); 128.0 (*d*); 149.1 (*s*); 169.8 (*s*). CI-MS: 311 ([*M* + 1]<sup>+</sup>).

*7. Quenching Experiment of the Photocycloaddition of 1a to Methacrylonitrile*. A soln. of **1a** (0.5 mmol), methacrylonitrile (8 mmol), and *trans*-stilbene (2 mmol) in benzene (40 ml) was irradiated at 366 nm, which was isolated with a filter soln. of naphthalene (3 g/500 ml MeOH). The photocycloaddition was completely quenched.

## REFERENCES

- [1] O. Buchardt, 'Photochemistry of Heterocyclic Compounds', Wiley, New York, 1976, p. 233.
- [2] I. Saito, S. Ito, *J. Synth. Org. Chem. Jpn.* **1978**, *36*, 1009.
- [3] T. Nishio, Y. Omote, *J. Chem. Soc., Perkin Trans. 1* **1988**, 957, and ref. cit. therein.
- [4] K.-H. Pfoertner, *Helv. Chim. Acta* **1975**, *58*, 865; Y. Kanaoka, M. Hasebe, Y. Hatanaka, *Heterocycles* **1979**, *13*, 263.
- [5] T. Takahashi, S. Hirokami, M. Nagata, T. Yamazaki, T. Date, *J. Chem. Soc., Perkin Trans. 1* **1989**, 1231, and ref. cit. therein.
- [6] a) T. Nishio, N. Nakajima, M. Kondo, Y. Omote, M. Kaftory, *J. Chem. Soc., Perkin Trans. 1* **1984**, 391; b) T. Nishio, M. Kondo, Y. Omote, *ibid.* **1985**, 2494; c) T. Nishio, N. Tokunaga, M. Kondo, Y. Omote, *ibid.* **1988**, 2921.
- [7] H. Furrer, *Chem. Ber.* **1972**, *105*, 2790.
- [8] T. Tsuchiya, M. Hasebe, H. Arai, H. Igeta, *Chem. Pharm. Bull.* **1974**, *22*, 2276; E. Sato, M. Hasebe, Y. Kanaoka, *Heterocycles* **1988**, *27*, 1665.
- [9] a) T. Nishio, *J. Org. Chem.* **1984**, *49*, 827; b) T. Nishio, Y. Omote, *ibid.* **1985**, *50*, 1370; c) T. Nishio, Y. Omote, *J. Chem. Soc., Perkin Trans. 1* **1987**, 2611; d) T. Nishio, *ibid.* **1990**, 565.
- [10] E. J. Corey, J. D. Baas, R. LeMaticu, R. B. Mitra, *J. Am. Chem. Soc.* **1964**, *86*, 5570.
- [11] P. S. Mariano, 'Photochemistry', Ed. A. Padwa, Marcel Dekker, New York, 1987, Vol. 9, p. 1.
- [12] E. Brill, H. P. Schultz, *J. Org. Chem.* **1964**, *29*, 579.